

M.Sc. Semester-III
Core Course - 7 (CC-7)
Application of Spectroscopy



III. Nuclear Magnetic Resonance Spectroscopy

L6: Coupling Constant, Stereochemical Nonequivalence

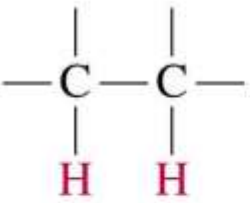
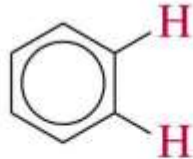
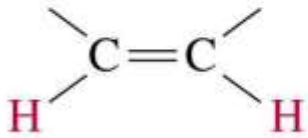
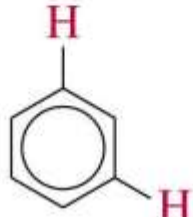
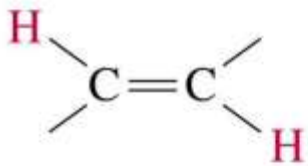
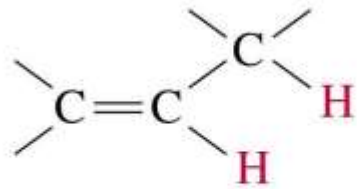
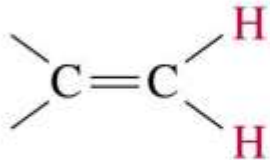


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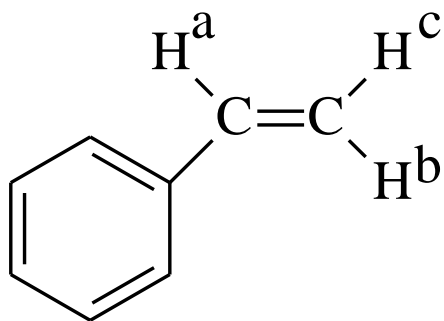
Coupling Constants

- Distance between the peaks of multiplet
- Measured in Hz
- Not dependent on strength of the external field
- Multiplets with the same coupling constants may come from adjacent groups of protons that split each other.

Values for Coupling Constants

		<u>Approx. J</u>			<u>Approx. J</u>
	(free rotation)	7 Hz ^a			8 Hz
	(cis)	10 Hz			2 Hz
	(trans)	15 Hz			6 Hz
	(geminal)	2 Hz		(allylic)	

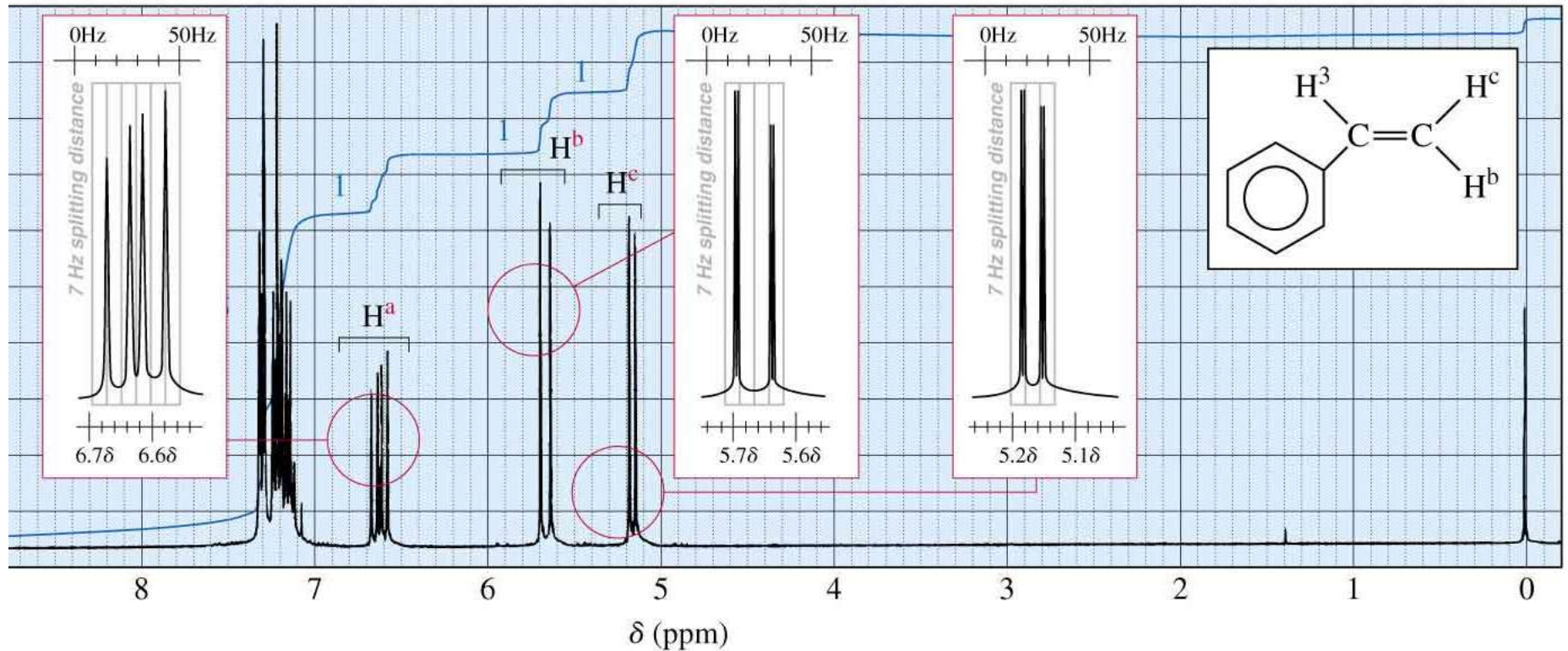
^aThe value of 7 Hz in an alkyl group is averaged for rapid rotation about the carbon-carbon bond. If rotation is hindered by a ring or bulky groups, other splitting constants may be observed.



Complex Splitting

- Signals may be split by adjacent protons, different from each other, with different coupling constants.
- Example: H^a of styrene which is split by an adjacent H *trans* to it ($J = 17$ Hz) and an adjacent H *cis* to it ($J = 11$ Hz).

Spectrum for Styrene

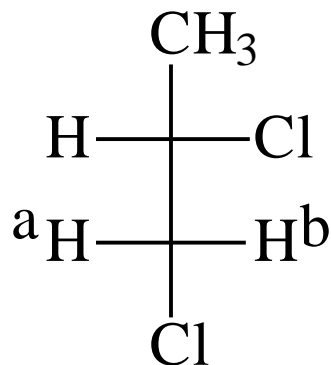
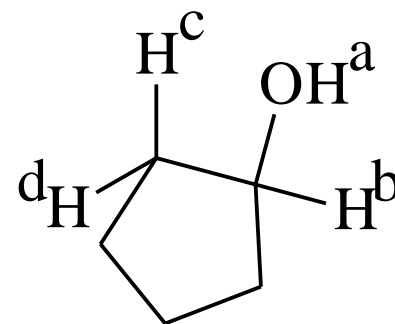
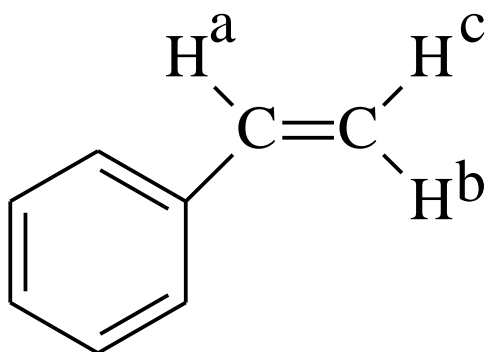


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Stereochemical Nonequivalence

- Usually, two protons on the same C are equivalent and do not split each other.
- If the replacement of each of the protons of a $-\text{CH}_2$ group with an imaginary “Z” gives stereoisomers, then the protons are non-equivalent and will split each other.

Some Nonequivalent Protons



Thank You



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